



Introductory Chemistry I (CHMA10H) Fall 2017 Syllabus University of Toronto at Scarborough

Dear Students:

Welcome to Introductory Chemistry Part I! Our primary goal is to ignite your passion for chemistry by creating a meaningful learning environment with many real life applications of chemistry. The knowledge you gain in this course is applicable in diverse disciplines, including Medicine, Pharmacy, Environmental Sciences, Neuroscience, Biochemistry and Biology. We are looking forward to teaching you many interesting topics on molecular structure, chemical reactions and nuclear chemistry. Please read through the course syllabus to understand the learning expectations and assessment methods. Looking forward to meeting all of you!

Teaching Team:

Instructor:

Dr. Nirusha Thavarajah

Office Room Number: ENV544, Email: nirusha.thavarajah@utoronto.ca

Office Hours: Monday, Wednesday & Friday 10:30-11:30 am

Lab Manager:

Dr. Scott Ballantyne: Room Number SW155C, Email: sballant@utsc.utoronto.ca

Office Hours: Mon and Wed 10:30 am to 12:00 pm

Recommended Preparation: Grade 12 Chemistry and Grade 12 Advanced Functions or Grade 12 Calculus.

Lectures:

LEC01 Mondays, Wednesday and Fridays: 12:00-1:00 pm at AC 223,

LEC02 Mondays, Wednesday and Fridays: 1:00-2:00 pm at AC223

We strongly encourage you to attend all the lectures to engage in the participatory lessons!

Email Policy: Please use a valid “utoronto.ca” or “utsc.utoronto.ca” account for all CHMA10 correspondence. Emails sent from other accounts are frequently filtered out as spam and may not be received. When composing your email, please use professional language. Be sure to include the course code as part of the subject line and sign the email with your first and last name, as well as your student ID. Your email will be answered as soon as possible (likely within 36 hours, unless it is a weekend or holiday). **Please Note: Double check the course syllabus before emailing us with a question. Questions on the lab material should be directed to the lab coordinator or your TA.**

Textbook: Chemistry: A Molecular Approach, Second Canadian Edition Plus MasteringChemistry Access Card Package (2nd Edition) by Nivaldo J. Tro *et al.*

Website: CHMA10 maintains the Blackboard web space which archives a variety of course-related information. There are two instructors for the course and at the beginning of each of their sections, they will inform you where you will find: contact information, class announcements, lecture slides,

handouts, assigned readings, suggested end-of-chapter problems, and links to some useful outside resources.

Accessing WebOption Lecture Videos

WebOption lecture videos can be accessed at the Blackboard via the “Lecturecasts” link.

Announcements: Official announcements regarding test locations, material covered for each test and other important announcements will be posted on the CHMA10H3 course web site. It is absolutely your responsibility to check these postings regularly for important announcements.

Accessibility: Students with diverse learning styles and needs are welcome in this course. If you require accommodations for a disability, or have any accessibility concerns about the course, the classroom or course materials, please contact us and/or the Accessibility Services as soon as possible: SW 302, (416) 287-7560 or ability@utsc.utoronto.ca

Active Learning in Introductory Chemistry I:

Course Surveys: There will be a pre-class and a mid-class surveys during the term to gather student feedback. More information on this will be announced on blackboard.

Peer Facilitator Program: Facilitated Study Group (FSG) is being run through the Centre for Teaching and Learning. These weekly sessions are open to all students taking this course who want to improve their understanding of course material, improve their study techniques, and improve their grade. Attendance is voluntary. In these sessions you will compare notes, discuss important concepts, develop study strategies, and prepare for exams and assignments on course material. Course material is NOT re-lectured. The FSG’s are led by a trained facilitator who has previously taken the course. A survey will be taken during the first week of class to determine the best days and times for most students, and they will begin probably the 2nd or 3rd week of class.

Lecture Topics: Below is a brief list of topics that will be covered in this course, along with the corresponding chapters. A more detailed list, with the associated textbook readings and assigned end-of-chapter problems will be posted on portal under course materials.

- Chemical Reactions and Stoichiometry (Chapter 4)
- Gases (Chapter 5)
- Thermochemistry (Chapter 6)
- Liquids, Solids and Intermolecular forces (Chapter 11)
- Solutions (Chapter 12)
- The Quantum-Mechanical Model of the Atom (Chapter 7)
- Periodic Properties of the Elements (Chapter 8)
- Chemical Bonding I: Lewis Theory (Chapters 9)
- Chemical Bonding II: (Molecular Shapes, Valence Bond Theory and Molecular Orbital Theory (Chapter 10)
- Radioactivity and Nuclear Chemistry (Chapter 19)
- Parts of Chapters 21-22, & 26 (If time permits, optional)

Assessment and Grading Practices:

Methods of Evaluation	Contribution to the Final Grade
Problem Sets (more details to be provided on Course web space on Blackboard)	5%
Laboratory* (<i>Must pass the lab component to pass the course</i>)	25%
Mid-Term Test (<i>*No-Make Up</i>) *If you miss a test with a legitimate reason supported by acceptable documentation, its value will be added to that of the final.	25%
Final Exam	45%
TOTAL	100%

Note: To pass the course, you **MUST** pass the laboratory and either the term test or the final exam (and receive a final grade of 50+, of course!).

Term Test – 25% of final grade: There will be a 100-minute term test around the middle to end of October which will count as 25% of your final grade. This test will be written outside of class time. The exact date, time and location will be announced as soon as this information is made available from the registrar.

Policy on Missed Tests: *This course will not have any make-up tests.* Should you miss the term test due to a legitimate reason, you must submit appropriate documentation *within one week of your absence*. If the reason is medical, an official UTSC medical form should be downloaded from http://www.utsc.utoronto.ca/~registrar/resources/pdf_general/UTSCmedicalcertificate.pdf and completed by your doctor. *If no acceptable documentation is received, you will receive a grade of zero for that test.*

With a validated absence, the value of the missed test will be added to your final exam. Please note that in UTSC Calendar it states: "You cannot petition to withdraw from a course on the grounds that no work was returned to you before the last day to withdraw without academic penalty if this is the result of your having been given an extension to complete your work for reasons relating to you and not the rest of your class."

Final Examination – 45% of final grade: There will be a 3-hour, *cumulative* exam written during the end of semester exam period. The exact date, time and location will be announced as soon as they are available. *Please note that if you miss the Final Exam, you must petition the Registrar's Office to write a make-up exam in the next formal exam period.* Check the UTSC Calendar for instructions and deadlines.

Online Grades: Individual grades will be posted on the Blackboard as they become available. Please check these periodically to make sure that the posted grades match your own records. Any discrepancy should be reported immediately to the instructor or the lab coordinator, as appropriate.

Labs: The laboratory component of CHMA10 is compulsory. **In order to pass the course, you **MUST** also pass the lab component.**

Lab Schedule:

Laboratory periods are three hours in length and run every other week. Odd numbered practical's (Week 1 students) have their first during week of September 11th. Even numbered practical's (Week 2 students) will have their first lab the week of September 18th.

Lab Manual and Notebook: A lab manual must be purchased from the UTSC Bookstore before your first lab. You may not use a lab manual from a previous semester: the experiments and course requirements will be different.

DO NOT wait to purchase your lab manual as it contains a host of important information:

- Lab Schedules and other important dates
- Late and absence policies
- Rules regarding safety
- Appropriate attire for the labs
- Guidelines on how to properly prepare for the lab

The bookstore **DOES NOT** stock enough lab manuals for everyone. If they run out, you **MUST** preorder a copy through the bookstore – this takes time. Failure to adhere to the rules and policies outlined within the lab manual will adversely affect your lab mark – in some instances the impact will be severe. In addition, students will be required to purchase their own lab notebook. The book must be hard-cover, permanently bound (not spiral or loose leaf) with the approximate dimensions 8.25” x 10.5” inches. They can be purchased at the UTSC bookstore; however students are free to purchase their books at a merchant of their choice (so long as they meet the above requirements).

Lab Safety: Safety in the laboratory is an extremely important element in the chemistry program at this University. Failure to follow safe practices can cause laboratory accidents which may result in the loss of time, damage to clothing and other property, and most importantly personal injury. By following suitable precautions, you can anticipate and prevent situations that would otherwise lead to accidents. Students registered in CHMA10H3F will be automatically enrolled in the WHMIS 2015 Training course for the Fall 2017 semester

Once the course is made available an email announcement will be made and a link to the course will appear in your “My Courses” window of your blackboard home page. As part of this course, students will be expected to watch a couple of videos (approximately 90 minutes long in total) and take a multiple choice quiz on the material you just learned. Students must obtain 80% on the quiz to pass the WHMIS course. **In addition, students will be required to print off your quiz results and present them to your TA before you will be allowed to enter the lab.**

Safety Equipment: Students will be required to purchase approved indirect vented chemical splash safety goggles, and a lab coat before attending their first lab. These items can be purchased from both the Environmental and Physical Sciences Student Association (EPSA) and the Biology Student Association (BioSA) or through the bookstore. All safety eyewear must meet either ANSI Z87+ or CSA Z94.3 Standard for high impact protection (if you see one of those standards stamped on your eyewear somewhere then they meet that particular standard). As part of your ancillary fees, all CHMA10H3F students will be provided a pair safety glasses at their first lab session which can be

worn during your quizzes and pre-lab discussion; however when the experiment begins, students will be required to wear their indirect vented chemical splash goggles.

Labs coats must be 100% cotton – no exceptions.

Further information regarding appropriate attire please see the guidelines outlined in your lab manual.

Note that students not wearing approved safety gear will not be allowed to participate in the lab.

Ancillary Fees: Students taking CHMA10 will be assessed a \$20.00 ancillary fee which will cover the cost of chemicals, filter paper, Pasteur pipettes and other items consumed over the course of the lab. For more information regarding ancillary fees students are encouraged to visit the following website:
http://www.planningandbudget.utoronto.ca/tuition/Ancillary_Fees.htm

Aids Allowed in Lab quizzes & Tests, Midterm Test and Final Exams:

Please note** that only non-programmable, non-communicating calculators are allowed in tests and exams for this course (both lecture and lab). Invigilators have the authority to check calculators during the tests and exams. Students who have illegal calculators confiscated during a test/exam will be supplied with an allowed calculator but an immediate penalty of 10% will be imposed for that test/exam. Students without a calculator will also be allowed to borrow an allowed model, but at the cost 10% off their mark on that test/exam.

Academic Integrity: Academic integrity is one of the cornerstones of the University of Toronto. It is critically important both to maintain our community which honours the values of honesty, trust, respect, fairness and responsibility and to protect you, the students within this community, and the value of the degree towards which you are all working so diligently. Detailed information about how to act with academic integrity, the Code of Behaviour on Academic Matters, and the processes by which allegations of academic misconduct are resolved can be found online:
<http://www.artsci.utoronto.ca/osai/students>

According to Section B of the University of Toronto's Code of Behaviour on Academic Matters <http://www.governingcouncil.utoronto.ca/policies/behaveac.htm> which all students are expected to know and respect, it is an offence for students to:

- To use someone else's ideas or words in their own work without acknowledging that those ideas/words are not their own with a citation and quotation marks, i.e. to commit plagiarism.
- To include false, misleading or concocted citations in their work.
- To obtain unauthorized assistance on any assignment.
- To provide unauthorized assistance to another student. This includes showing another student completed work.
- To submit their own work for credit in more than one course without the permission of the instructor.
- To falsify or alter any documentation required by the University. This includes, but is not limited to, doctor's notes.
- To use or possess an unauthorized aid in any test or exam.

There are other offences covered under the Code, but these are by far the most common. Please respect these rules and the values which they protect. Offences against academic integrity will be dealt with according to the procedures outlined in the Code of Behaviour on Academic Matters.